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# REPORT FOR COMPLIANCE TESTING

Detroit Refinery KHT Charge Heater Stack

Marathon Petroleum Company LP 1300 South Fort Street Detroit, MI 48217 Client Reference No. 4101004604 CleanAir Project No. 13265-1
STAC Certificate No. 2007.002.0113.1217
Revision 0, Final Report
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## 1. PROJECT OVERVIEW

## Test Program Summary

Marathon Petroleum Company LP (MPC) contracted CleanAir Engineering (CleanAir) to successfully complete testing on the KHT Charge Heater (EU19-KHTCHARHTR-S1) at the Detroit Refinery located in Detroit, Michigan. The objective of the test program was to perform particulate matter (PM), total particulate matter less than 10 microns in diameter (PM $_{10}$ ), sulfuric acid (H $_2$ SO $_4$ ), nitrogen oxides (NO $_X$ ) and carbon monoxide (CO) testing to demonstrate compliance with the Michigan Department of Environmental Quality (MDEQ) Permit No. MI-ROP-A9831-2012c.

A summary of the test program results is presented below. Section 2 Results provides a more detailed account of the test conditions and data analysis. Test program information, including the test parameters, on-site schedule and a project discussion, begin on page 2.

Table 1-1: Summary of Results

Source Constituent	Sampling Method	Average Emission	Permit Limit <sup>1</sup>
KHT Charge Heater			
PM (lb/MMBtu)	USEPA5	0.0013	0.0019
PM <sub>10</sub> (lb/MMBtu)	USEPA 5/202	0.0028	0.0076
H <sub>2</sub> SO <sub>4</sub> (Ib/MMBtu)	Draft ASTM CCM	8.2E-04	N/A
NOx (Ib/MMBtu)	USEPA 7E	0.15	0.20
CO (lb/MMBtu)	USEPA 10	<0.0004	0.02

<sup>&</sup>lt;sup>1</sup> Permit limits obtained from MDEQ Permit No. MI-ROP-A9831-2012c.

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# Test Program Details

#### **Parameters**

The test program included the following emissions measurements:

- particulate matter (PM), assumed equivalent to filterable particulate matter (FPM) only
- total particulate matter less than 10 microns in diameter (PM<sub>10</sub>), assumed equivalent to the sum of the following constituents:
  - o filterable particulate matter (FPM)
  - o condensable particulate matter (CPM)
- nitrogen oxides (NO<sub>x</sub>)
- carbon monoxide (CO)
- sulfuric acid mist (H<sub>2</sub>SO<sub>4</sub>)
- flue gas composition (e.g., O<sub>2</sub>, CO<sub>2</sub>, H<sub>2</sub>O)
- flue gas temperature
- · flue gas flow rate

### Schedule

Testing was performed on May 22 and 23, 2017. The on-site schedule followed during the test program is outlined in Table 1-2.

Table 1-2: Test Schedule

Run Number	Location	Method	Analyte	Date	Start Time	End Time
1	KHT Charge Heater	USEPA Method 5/202	FPWCPM	05/22/17	09:40	11:46
2	KHT Charge Heater	USEPA Method 5/202	FPM/CPM	05/22/17	12:18	14:21
3	KHT Charge Heater	USEPA Method 5/202	FPM/CPM	05/22/17	16:38	18:42
1	KHT Charge Heater	USEPA Method 3A/7E/10	O <sub>2</sub> /CO <sub>2</sub> /NO <sub>x</sub> /CO	05/22/17	10:01	11:06
2	KHT Charge Heater	USEPA Method 3A/7E/10	O <sub>2</sub> /CO <sub>2</sub> /NO <sub>x</sub> /CO	05/22/17	11:32	12:32
3	KHT Charge Heater	USEPA Method 3A/7E/10	O <sub>2</sub> /CO <sub>2</sub> /NO <sub>x</sub> /CO	05/22/17	12:57	14:05
0	KHT Charge Heater	Draft ASTM CCM	Sulfuric Acid	05/23/17	09:54	10:54
1	KHT Charge Heater	Draft ASTM CCM	Sulfuric Acid	05/23/17	11:09	12:09
2	KHT Charge Heater	Draft ASTM CCM	Sulfuric Acid	05/23/17	12:38	13:38
3	KHT Charge Heater	Draft ASTM CCM	Sulfuric Acid	05/23/17	13:55	14:55

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#### Discussion

Test Scope Synopsis

#### FPM & PM<sub>10</sub> Testing

A total of three (3) valid 120-minute Method 5/202 test runs were performed. FPM/CPM emission results were calculated in units of pounds per million Btu (lb/MMBtu). The final result was expressed as the average of the three (3) valid runs.

For this test program, PM emission rate is assumed equivalent to FPM. PM $_{10}$  is assumed equivalent to the sum of FPM less than 10 micrometers ( $\mu$ m) in diameter (FPM $_{10}$ ) and CPM. The Method 5/202 sample train yields a front-half, FPM result and a back-half, CPM result. The total PM result (FPM plus CPM) from Method 5/202 can be used as a worst-case estimation of total PM $_{10}$  since Method 5 collects all FPM present in the flue gas (regardless of particle size).

#### O2, CO2, NOX & CO Testing

Minute-average data points for O<sub>2</sub>, CO<sub>2</sub>, NO<sub>X</sub> and CO (dry basis) were collected over a period of 60 minutes for each run utilizing EPA Methods 3A, 7E and 10.

For all three (3) runs, the CO concentration was below 1% of the instrument calibration span. As such, CO emission rates are expressed as "less than" the emission rate calculated using an assumed concentration equivalent to 1% of the instrument calibration span.

#### H<sub>2</sub>SO<sub>4</sub> Testing

H<sub>2</sub>SO<sub>4</sub> emissions were determined referencing the Draft ASTM Controlled Condensation Method.

Three (3) 60-minute Draft ASTM CCM test runs were performed. H₂SO₄ emission results were calculated in units of lb/MMBtu. The final results were expressed as the average of three (3) valid runs.

Prior to the first official test run, a 60-minute sample conditioning run (Run 0) was performed in order to minimize the absorption capacity of the front-half components of the sample train (upstream of the  $H_2SO_4$ -collecting portion of the sample train). The conditioning run was recovered in the same manner as the official test runs, but the condenser rinse and SAM filter were not analyzed.

## Fuel Analysis

Emission results in units of dry volume-based concentration (lb/dscf, ppmdv) were converted into units of pound per million BTU (lb/MMBtu) by calculating an oxygen-based fuel factor ( $F_d$ ) for refinery gas per USEPA Method 19 specifications. The  $F_d$  factor was calculated from percent volume composition analytical data provided by MPC and tabulated heating values for each of the measured constituents.

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#### Test Conditions

The unit was operated above 8,200 barrels/day (bpd) during  $NO_X$  and CO emissions compliance testing, and during Runs 1 and 2 of particulate emission testing. During  $H_2SO_4$  emission testing and Run 3 of particulate emission testing, the unit was operated at 5,600-5,700 bpd. MPC was responsible for logging any relevant process-related data and providing it to CleanAir for inclusion in this test report.

End of Section

# 2. RESULTS

This section summarizes the test program results. Additional results are available in the report appendices, specifically Appendix C Parameters.

Table 2-1: KHT Charge Heater Stack – FPM & PM<sub>10</sub> Emissions

Run No.	•	1	2	3	Average
Date (20	017)	May 22	May 22	May 22	
Start Time (approx.)		09:40	12:18	16:38	
Stop Tir	me (approx.)	11:46	14:21	18:42	
Proces	s Conditions				
P <sub>1</sub>	Fuel gas flow rate (Mscf/day)	340	329	214	294
$P_2$	Charge rate (bpd)	8,308	8,077	5,655	7,347
$F_d$	Oxygen-based F-factor (dscf/MMBtu)	8,315	8,315	8,315	8,315
Gas Coi	nditions				
$O_2$	Oxygen (dry volume %)	7.0	7.3	8.3	7.5
$CO_2$	Carbon dioxide (dry volume %)	8.1	7.9	7.3	7.8
$T_{s}$	Sample temperature (°F)	816	810	678	768
$B_w$	Actual water vapor in gas (% by volume)	13.3	12.7	12.7	12.9
Gas Flor	w Rate				
$Q_a$	Volumetric flow rate, actual (acfm)	10,100	10,100	6,600	8,930
$Q_s$	Volumetric flow rate, standard (scfm)	4,160	4,150	3,040	3,780
$\mathbf{Q}_{\mathrm{std}}$	Volumetric flow rate, dry standard (dscfm)	3,610	3,620	2,650	3,290
Samplin	ng Data		•	•	•
$V_{mstd}$	Volume metered, standard (dscf)	73.41	73.71	82.88	76.67
<b>%</b> I	Isokinetic sampling (%)	100.6	100.6	102.2	101.1
Laborat	tory Data		•		
$m_{FPM}$	Total FPM(g)	0.00421	0.00364	0.00269	
m <sub>CPM</sub>	Total CPM(g)	0.00400	0.00412	0.00370	
m <sub>Part</sub>	Total particulate matter (as PM <sub>10</sub> ) (g)	0.00821	0.00776	0.00639	•
FPM Res	sults				
$C_{sd}$	Particulate Concentration (lb/dscf)	1.26E-07	1.09E-07	7.16E-08	1.02E-07
E <sub>lb/hr</sub>	Particulate Rate (lb/hr)	0.0274	0.0237	0.0114	0.0208
$E_{Fd}$	Particulate Rate - F₀ based (lb/MMBtu)	0.00158	0.00139	0.00099	0.00132
CPM Re	sults				
$C_{\rm sd}$	Particulate Concentration (lb/dscf)	1.20E-07	1.23E-07	9.85E-08	1.14E-07
E <sub>lb/hr</sub>	Particulate Rate (lb/hr)	0.0260	0.0268	0.0157	0.0228
$E_{Fd}$	Particulate Rate - F <sub>d</sub> -based (lb/MMBtu)	0.00150	0.00158	0.00136	0.00148
Total Pa	articulate Matter (as PM <sub>10</sub> ) Results				
$C_{sd}$	Particulate Concentration (lb/dscf)	2.47E-07	2.32E-07	1.70E-07	2.16E-07
E <sub>lb/b</sub>	Particulate Rate (lb/hr)	0.0534	0.0505	0.0271	0.0436
$E_{Fd}$	Particulate Rate - F <sub>d</sub> -based (lb/MMBtu)	0.00308	0.00297	0.00235	0.00280

Table 2-2: KHT Charge Heater Stack – CO & NO<sub>X</sub> Emissions

Run No.		1	2	3	Average
Date (2017)		May 22	May 22	May 22	
Start Tin	ne (approx.)	10:01	11:32	12:57	
Stop Tin	ne (approx.)	11:06	12:32	14:05	
Process	s Conditions				
$P_1$	Fuel gas flow rate (Mscf/day)	339	341	331	337
$P_2$	Charge rate (bpd)	8,308	8,301	8,079	8,229
$F_{d}$	Oxygen-based F-factor (dscf/MMBtu)	8,315	8,315	8,315	8,315
Gas Cor	nditions				
O <sub>2</sub>	Oxygen (dry volume %)	6.3	6.4	6.7	6.5
$CO_2$	Carbon dioxide (dry volume %)	8.7	8.7	8.5	8.6
Carbon	Monoxide Results				
$C_{sd}$	Concentration (ppmdv) <sup>1</sup>	< 0.479	< 0.479	<0.479	< 0.479
$C_{\sf sd}$	Concentration (lb/dscf)	<3.48E-08	<3.48E-08	<3.48E-08	<3.48E-08
$E_{Fd}$	Emission Rate - F <sub>d</sub> -based (lb/MMBtu)	<4.15E-04	<4.17E-04	<4.26E-04	<4.20E-04
Nitroger	n Oxides Results				
$C_{sd}$	Concentration (ppmdv)	107	107	107	107
$C_{\rm sd}$	Concentration (lb/dscf)	1.28E-05	1.28E-05	1.28E-05	1.28E-05
$E_{Fd}$	Emission Rate - F <sub>d</sub> -based (lb/MMBtu)	0.152	0.154	0.157	0.154

<sup>&</sup>lt;sup>1</sup>Resultant concentrations were less than 1% of instrument calibration span.

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Table 2-3: KHT Charge Heater Stack -- H<sub>2</sub>SO<sub>4</sub> Emissions

Run No	).	1	2	3	Average
Date (2017)		May 23	May 23	May 23	
Start Ti	me (approx.)	11:09	12:38	13:55	
Stop Ti	me (approx.)	12:09	13:38	14:55	
Proces	s Conditions				
$P_1$	Fuel gas flow rate (Mscf/day)	197	195	196	196
$P_2$	Charge rate (bpd)	5,646	5,625	5,608	5,626
$F_{d}$	Oxygen-based F-factor (dscf/MMBtu)	8,337	8,337	8,337	8,337
Gas Co	nditions				
$O_2$	Oxygen (dry volume %)	7.0	7.1	6.6	6.9
$CO_2$	Carbon dioxide (dry volume %)	8.0	7.9	8.2	8.0
$T_{s}$	Sample temperature (°F)	588	590	584	587
$B_w$	Actual water vapor in gas (% by volume)	13.4	13.5	14.1	13.7
Sampli	ng Data				
V <sub>mstd</sub>	Volume metered, standard (dscf)	28.19	27.81	27.81	27.94
Labora	tory Data (lon Chromatography)				
$m_n$	Total H2SO4 collected (mg)	0.7425	0.8560	0.9088	
Sulfurio	c Acid Vapor (H₂SO₄) Results				
$C_{\sf sd}$	H <sub>2</sub> SO <sub>4</sub> Concentration (lb/dscf)	5.81E-08	6.79E-08	7.21E-08	6.60E-08
$C_{sd}$	H <sub>2</sub> SO <sub>4</sub> Concentration (ppmdv)	0.228	0.267	0.283	0.259
$E_{Fd}$	H <sub>2</sub> SO <sub>4</sub> Rate - Fd-based (lb/MMBtu)	7.28E-04	8.57E-04	8.78E-04	8.21E-04

End of Section

# 3. DESCRIPTION OF INSTALLATION

## **Process Description**

MPC's facility in Detroit, Michigan, produces refined petroleum products from crude oil. MPC must continue to demonstrate that select process units are in compliance with permitted emission limits.

The Kerosene Hydrotreater unit uses hydrogen to remove sulfur and nitrogen from kerosene (and occasionally Naphtha). This process, known as hydrotreating, uses a catalyst with hydrogen and temperature to promote the desulfurization reaction. The KHT unit consists of process vessels (including exchangers, a reactor, a receiver, separators and a stripper column), a heater, tanks, containers, pumps, compressors, piping, drains and various components (pump and compressors, seals, process valves, pressure relief valves, flanges, connectors, etc.).

The KHT Heater (EU19-KHTCHARHTR-S1) preheats the feed to the reactor. The unit is fired by refinery fuel gas. Emissions are vented to the atmosphere via the KHT Charge Heater Stack (SV19-H2) where testing was performed.

#### Test Location

The sample point locations were determined by EPA Methods 1 and 7E specifications. Table 3-1 presents the sampling information for the test location described in this report. The figures shown on pages 9 and 10 represent the layout of the test location.

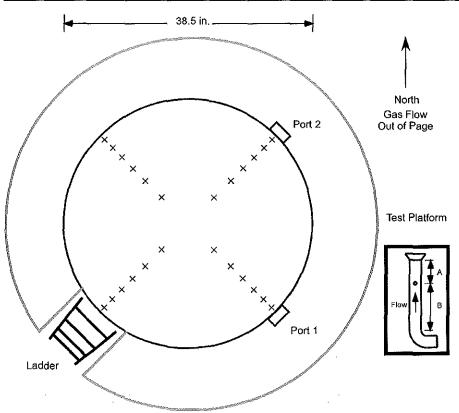
Table 3-1: Sampling Point Information

Source		Run	_	Points per	Minutes	Total	_
Constituent	Method	No.	Ports	Port	per Point	Minutes	Figure
KHT Charge Heater							
FPM/CPM (PM <sub>10</sub> )	EPA M5/202	1-3	2	12	5	120	3-1
H <sub>2</sub> SO <sub>4</sub>	Draft ASTM CCM	1-3	1	1	60	60	N/A <sup>1</sup>
$O_2/CO_2/NO_X/CO$	EPA M3A/7E/10	1-3	1	1 <sup>2</sup>	60	60	3-2

<sup>&</sup>lt;sup>1</sup> Sampling occurred at a single point near the center of the duct.

<sup>&</sup>lt;sup>2</sup> A stratification check conducted prior to Run 1 determined source met criteria for single point sampling.

Figure 3-1: FPM & PM<sub>10</sub> Sample Point Layout (EPA Method 1)



Sampling Point	% of Stack Diameter	Port to Point Distance (inches)
1	97.9	37.5
2	93.3	35.9
3	88.2	34.0
4	82.3	31.7
5	75.0	28.9
6	64.4	24.8
7	35.6	13.7
8	25.0	9.6
9	17.7	6.8
10	11.8	4.5
11	6.7	2.6
12	2.1	1.0

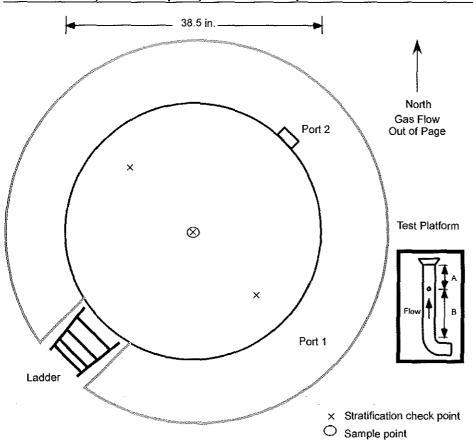
Duct diameters upstream from flow disturbance (A): 5.0 Duct diameters downstream from flow disturbance (B): 3.7

Limit: 0.5

Limit: 2.0

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Figure 3-2: CO & NO<sub>X</sub> Sample Point Layout (EPA Method 7E)



Sampling Point	% of Stack Diameter	Port to Point Distance (inches)
1	83.3	32.1
2	50.0	19.3
3	16.7	6.4

Duct diameters upstream from flow disturbance (A): 5.0 Duct diameters downstream from flow disturbance (B): 3.7 Limit: 0.5

Limit: 2.0

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## 4. METHODOLOGY

## Procedures and Regulations

The test program sampling measurements followed procedures and regulations outlined by the USEPA and the MDEQ. These methods appear in detail in Title 40 of the CFR and at https://www.epa.gov/emc. Appendix A includes diagrams of the sampling apparatus, as well as specifications for sampling, recovery and analytical procedures.

CleanAir follows specific QA/QC procedures outlined in the individual methods and in USEPA "Quality Assurance Handbook for Air Pollution Measurement Systems: Volume III Stationary Source-Specific Methods," EPA/600/R-94/038C. Appendix D contains additional QA/QC measures, as outlined in CleanAir's internal Quality Manual.

## Title 40 CFR Part 60, Appendix A

little 40 CF	R Part 60, Appendix A
Method 1	"Sample and Velocity Traverses for Stationary Sources"
Method 2	"Determination of Stack Gas Velocity and Volumetric Flow Rate (Type S Pitot Tube)"
Method 3	"Gas Analysis for the Determination of Dry Molecular Weight"
Method 3A	"Determination of Oxygen and Carbon Dioxide Concentrations in Emissions from Stationary Sources (Instrumental Analyzer Procedure)"
Method 3B	"Gas Analysis for the Determination of Emission Rate Correction Factor or Excess Air"
Method 4	"Determination of Moisture Content in Stack Gases"
Method 5	"Determination of Particulate Matter Emissions from Stationary Sources"
Method 7E	"Determination of Nitrogen Oxide Emissions from Stationary Sources (Instrumental Analyzer Procedure)"
Method 10	"Determination of Carbon Monoxide Emissions from Stationary Sources (Instrumental Analyzer Procedure)"
Method 19	"Determination of Sulfur Dioxide Removal Efficiency and Particulate Matter, Sulfur Dioxide and Nitrogen Oxide Emission Rates"

## Title 40 CFR Part 51, Appendix M

Method 202 "Dry Impinger Method for Determining Condensable Particulate Emissions from Stationary Sources"

# CTM-013 (Mod.)/Draft ASTM Controlled Condensation Method (Draft ASTM CCM)

"Determination of Sulfur Oxides Including Sulfur Dioxide, Sulfur Trioxide and Sulfuric Acid Vapor and Mist from Stationary Sources Using a Controlled Condensation Sampling Apparatus"

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## Methodology Discussion

#### FPM and PM<sub>10</sub> Testing – USEPA Method 5/202

The front-half (Method 5 portion) of the sampling train consisted of a glass nozzle, glass liner and filter holder heated to  $248^{\circ}F \pm 25^{\circ}F$  and a quartz fiber filter. Flue gas samples were extracted isokinetically per Method 5 requirements.

The back-half (Method 202 portion) of the sampling train is designed to mimic ambient conditions and collect only the particles that would truly form CPM in the atmosphere by minimizing the sulfur dioxide ( $SO_2$ ) and  $NO_X$  interferences observed with earlier versions of the method, in which flue gas was bubbled through cold water and  $SO_2$  and  $NO_X$  were absorbed and partially oxidized before they could be purged out with nitrogen ( $N_2$ ).

Flue gas exiting the front-half heated filter passed through a coiled condenser and dry impinger system jacketed by water continually circulated at ambient temperature. Moisture was removed from the flue gas without bubbling through the condensed water. Flue gas then passed through a tetrafluoroethane (TFE) membrane filter at ambient temperature. The temperature of the flue gas at the exit of the filter was directly measured with an in-line thermocouple and maintained in the temperature range of 65°F to 85°F.

After exiting the ambient filter, the flue gas passed through two (2) additional impingers surrounded by ice in a "cold" section of the impinger bucket. The moisture collected in these impingers were not analyzed for CPM and was only collected to determine the flue gas moisture and thoroughly dry the gas. The sample gas then flowed into a calibrated dry gas meter where the collected sample gas volume was determined.

The front-half portion of the sample train (nozzle, probe and heated filter) was recovered per Method 5 requirements, using acetone as the recovery solvent. The back-half of the sample train (heated filter outlet, condenser, dry impingers and TFE membrane filter) was recovered per Method 202 requirements. The impinger train was purged with nitrogen ( $N_2$ ) at a rate of 14 liters per minute (lpm) for one (1) hour following each test run and prior to recovery.

A field train blank was assembled, purged and recovered as if it were an actual test sample; analysis of the field train blank was used to blank-correct the test run results. Reagent blanks were also collected to quantify background contamination. All samples and blanks were returned to CleanAir Analytical Services for gravimetric analysis. Method 202 samples were maintained at a temperature < 85°F during transport to the laboratory.

## $O_2$ , $CO_2$ , $NO_X$ and CO Testing – USEPA Methods 3A and 10

Reference method oxygen  $(O_2)$  and carbon dioxide  $(CO_2)$  emissions were determined using a paramagnetic/NDIR analyzer per EPA Method 3A. Reference method CO emissions were determined using an infrared analyzer per EPA Method 10. Reference method  $NO_X$  emissions were determined using a chemiluminescent analyzer per EPA Method 7E.

Sample gas was extracted at a constant rate, conditioned to remove moisture and delivered to an analyzer bank which measured concentration on a dry basis (units of %dv or ppmdv).

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Calibration error checks were performed by introducing zero N<sub>2</sub>, high range and mid-range calibration gases to the inlet of each analyzer during calibration error checks. Bias checks were performed before and after each sampling run by introducing calibration gas to the inlet of the sampling system's heated filter. Per Methods 3A and 10, the average results for each run were drift-corrected.

## H<sub>2</sub>SO<sub>4</sub> Testing – Draft ASTM CCM

A gas sample was extracted from the source at a constant flow rate from the source using a quartz-lined probe maintained at a temperature of 650°F ± 25°F (depending on the required probe length) and a quartz fiber filter maintained at the same temperature as the probe to remove particulate matter.

The sample then passed through a glass coil condenser for collection of sulfuric acid vapor and/or mist. A second quartz fiber filter (referred to as the sulfuric acid mist (SAM) filter) was located at the condenser outlet for the collection of residual SAM not collected by the condenser. The condenser temperature was regulated by a water jacket and the SAM filter was regulated by a closed oven. Both the water jacket and SAM filter oven were maintained at  $140^{\circ}$ F  $\pm 9^{\circ}$ F.

After exiting the SAM filter, the sample gas continued through a series of four (4) glass knock-out jars; two (2) containing water, one (1) empty and one (1) containing silica gel for residual moisture removal. The exit temperature from the knock-out jar set was maintained below 68°F. The sample gas then flowed into a dry gas meter, where the collected sample gas volume was determined by means of a calibrated, dry gas meter or an orifice-based flow meter.

The  $H_2SO_4$ -collecting portion of the sample train (condenser and SAM filter) was recovered into a single fraction using DI  $H_2O$  as the recovery/extraction solvent; any  $H_2SO_4$  disassociates into sulfate ion ( $SO_4^{2-}$ ) and is stabilized in the  $H_2O$  matrix until analysis.

Samples and blanks were returned to CleanAir Analytical Services for ion chromatography (IC) analysis.

End of Section